

Expedient Synthesis of Highly Substituted Pyrroles via Tandem Rearrangement of α -Diazo Oxime Ethers

Yaojia Jiang, Wei Chuen Chan, and Cheol-Min Park*

Division of Chemistry and Biological Chemistry, School of Physical and Mathematical Sciences, Nanyang Technological University, Singapore 637371, Singapore

S Supporting Information

ABSTRACT: An efficient rhodium-catalyzed synthesis of 2*H*-azirines and pyrroles has been developed. Novel rearrangement of α -oximino ketenes derived from α -diazo oxime ethers provides 2*H*-azirines bearing quaternary centers and allows for subsequent rearrangement to highly substituted pyrroles in excellent yields.

P yrroles¹ are among the most important heterocycles that are frequently found in natural products,² pharmaceuticals,³ and functional materials.⁴ These utilities continue to drive the interest in the development of new synthetic methods for pyrroles which include cycloaddition reactions,⁵ multicomponent coupling reactions,⁶ and transition metal-mediated reactions.⁷ Padwa and co-workers reported synthesis of pyrroles via rearrangement of 2-vinyl-2*H*-azirines.⁸ Inspired by the report, we aimed to develop a synthetic method for 2*H*-azirines and pyrroles based on novel rearrangement of α -dizao oxime ethers (eq 1).



2H-Azirines represent a highly valuable class of compounds found in natural products and synthetic intermediates.⁹ Derived from the high ring strain present in these smallest heterocycles, their unique reactivity allows for 2H-azirines to serve as a versatile source of nitrenes, electrophiles, dienophiles, and dipolarophiles in various reactions.¹⁰ These reactions lead to the development of efficient synthetic platforms for various nitrogen containing heterocycles including pyrroles, indoles, pyrazolo[1,5-a]pyridines, isoxazoles, and piperidines.¹¹ Despite their broad utility, the availability of synthetic methods for 2H-azirines is rather limited, primarily relying on the Neber reaction¹² and thermal/photochemical rearrangement of vinyl azides.¹³ The efficiency of these reactions is, however, highly variable and dependent on substrates due to the sensitive nature of 2H-azirines that promotes further activation and subsequent decomposition. Also, these reactions typically require stoichiometric reagents and harsh reaction conditions. With the tandem synthesis of pyrroles in mind, we first set out to develop catalytic and neutral conditions for 2H-azirines

bearing *quaternary centers* required for the subsequent rearrangement.

Wolff rearrangement, among the fundamental reactions of diazo compounds, has found broad applications in organic synthesis.¹⁴ Ketenes generated by metal-catalyzed or thermal/ photochemical activation of diazo compounds have proven as versatile intermediates in various transformations including nucleophilic addition, cycloaddition, ring contraction, and homologation.¹⁵ However, to the best of our knowledge, no examples of 2*H*-azirine synthesis via Wolff rearrangement have been reported. In conjunction with our recent studies on the chemistry of α -diazo oxime ethers,¹⁶ we investigated the feasibility of metal-mediated cascade rearrangement leading to formation of 2*H*-azirine-2-carboxylic esters (eq 2) and subsequent rearrangement to pyrroles by introduction of vinyl groups on α -diazo oxime ethers (eq 1).



Our attempts began with a reaction on α -diazo- β -keto oxime ether 1a employing Cu(OTf)₂ as catalyst (Scheme 1).¹⁷





^{*a*}OTf = trifluoromethanesulfonate, OBz = benzoate. Yields determined by NMR vs standard.

Thus, exposure of 1a to $Cu(OTf)_2$ in 1,2-dichloroethane (DCE) at 60 °C resulted in formation of two products in low yields that were identified as 2a and 2a'. Compound 2a appears to arise from the desired cascade rearrangement,

Received: January 21, 2012 Published: February 14, 2012 while 2a' results from N–O insertion.¹⁸ With the result in hand, we proceeded with screening of various metal salts (see the Supporting Information). Use of Ag(OBz) which is frequently employed in the Wolff rearrangement gave improvement in yield, while the selectivity between the cascade rearrangement and N–O insertion remained unsatisfactory. To our delight, exposure of 1a to 2 mol % of Rh₂(OAc)₄ led to exclusive formation of 2a in nearly quantitative yield.¹⁹ Examination of solvent effects on the rearrangement revealed that the reaction is well tolerated in various solvents (see the Supporting Information).

With the optimal conditions in hand, we next examined the substrate scope of 2*H*-azirine formation (Table 1). Consistent

Table 1. Substrate Scope^a



^{*a*}The reported yields in parentheses are of the isolated products. ^{*b*}Byproduct: 1-(2-methoxy-3-phenyl-2*H*-azirin-2-yl)butan-1-one 2*i*' (29%). ^{*c*}Byproducts: 1-(2-methoxy-3-phenyl-2*H*-azirin-2-yl)-2,2-dimethylpropan-1-one 2*k*' (46%); 2,2-dimethyl-1-(3-phenyl-2*H*-azirin-2yl)propan-1-one 2*k*'' (18%). ^{*d*}Byproduct: (*E*)-1-(2-methoxy-3-phenyl-2*H*-azirin-2-yl)hex-2-en-1-one 2*l*' (18%). ^{*e*}Byproduct: (2-methoxy-3phenyl-2*H*-azirin-2-yl)(phenyl)methanone 2*o*' (44%) obtained as an inseparable mixture with 2*o* (yields were determined by NMR). ^{*f*}Byproduct: (4-methoxyphenyl)(3-phenyl-2*H*-azirin-2-yl)methanone 2*p*'' (21%). ^{*g*}Byproduct: furan-2-yl(2-methoxy-3-phenyl-2*H*-azirin-2-yl).

with the results in the optimization studies, reactions smoothly proceeded to afford 2*H*-azirine-2-carboxylic esters in good to excellent yields. To probe electronic and steric influences on

the rearrangement, various types of substituents for each of the three sites (R^1 , R^2 , and R^3) were examined. For R^1 adjacent to the oxime ethers, alkyl groups were well tolerated (**2b** and **2c**). Also, substrates with alkenyl, alkynyl, and aryl substituents smoothly reacted to give the corresponding 2*H*-azirine-2-carboxylic esters in excellent yields (**2d**–**g**). Examination of the electronic effect of R^1 by varying substituents on phenyl groups revealed little impact on the reaction efficiency; 92% and 96% for 4-methoxyphenyl (**2f**) and 4-nitrophenyl (**2g**) compounds, respectively. Furthermore, the substrate with heteroaryl group such as 2-furyl group afforded **2h** in 87% yield.

Next, we turned our attention to the influence of R² undergoing migration (Table 1, 2i-q). Examination of various R^2 revealed the efficiency of the reaction in the order of secondary, vinyl \approx primary followed by tertiary alkyl groups: 2j, 94%; 21, 74%; 2i, 66%; 2k, 33%. For those with low yields, the major byproducts were the corresponding N-O insertion products (Table 1, footnote). For 1k, additional byproduct 2k" was isolated in 18% yield that results from 1,5-hydride shift from the methyl ether.^{20,21} To examine the potential isomerization of (E)- and (Z)-alkenes during migration, 11 and 1m were subjected to the reaction conditions. Gratifyingly, both substrates gave the corresponding isomers with complete retention of geometry (Table 1, 2l and 2m). Moreover, 1n with a cyclohexene moiety also produced 2n in 82% yield. Next, we examined the influence of migrating groups when $R^2 = aryl/$ heteroaryl groups. While the substrates with phenyl (10) and 2-furyl (1q) groups afforded the expected products 20 and 2q (44% and 60%, respectively) along with N-O insertion products 2o' and 2q' (44% and 31%, respectively), 1p bearing more electron-rich 4-methoxyphenyl group gave 2p in higher yield (70%) along with byproduct 2p" (21%) formed via 1,5hydride shift²⁰ (Table 1, footnote). A brief examination of the alkoxy group (OR³) of oxime ethers showed that both electrondeficient (1r) and phenyl (1s) groups are also well tolerated to afford the corresponding esters in excellent yields.

With the efficient synthetic protocol for 2H-azirines in hand, we explored the viability of tandem rearrangement to pyrroles (Table 2). The study by Padwa and co-workers demonstrated that 2-vinyl-2H-azirines undergo rearrangement to furnish pyrroles via nitrenes.⁸ We reasoned that the rhodium catalysis in the formation of 2-vinyl-2H-azirines may be further exploited in the synthesis of pyrroles in tandem fashion. We were delighted to find that exposure of α -diazo oxime ether 3a to 2 mol % $Rh_2(OAc)_4$ in refluxing toluene provided tetrasubstituted pyrrole 4a in 90% yield. Encouraged by this result, we examined the substrate scope of the reaction. As shown in Table 2, this protocol offers highly flexible synthesis of pyrroles with various substituents. While substrates with alky groups on vinyl substituents required refluxing toluene (4a, 4n, and 4l), those with any groups for \mathbb{R}^3 gave the corresponding pyrroles at 60 °C in excellent yields (4b-h).

Mechanistically, we propose that the rearrangement proceeds through ketene intermediate C. Thus, rhodium-catalyzed formation of carbenoid B promotes migration of the vinyl substitutent to the carbenoid center resulting in formation of ketene C. Attack of the oxygen atom of the oxime ether moiety on the ketene leads to formation of ylides D/E that further undergo rearrangement to afford 2*H*-azirine-2-carboxylic ester F. Rhodium–nitrene complex G formed via ring-opening of F undergoes C–H insertion to afford pyrrole H.

To probe the presence of ketenes as intermediates in the reaction pathway as shown in Scheme 2, we demonstrated that

Table 2. Synthesis of Pyrroles via Tandem Reaction of α -Diazo Oxime Ethers^{*a*}



^{*a*}The reported yields in parentheses are of the isolated products. ^{*b*}Toluene, 110 °C. ^{*c*}DCE, 60 °C. ^{*d*}(*E*)-3-(3-Cyclohexyl-2-methoxy-2*H*-azirin-2-yl)-1-phenylprop-2-en-1-one **4** h' (19%) obtained as byproduct.

Scheme 2. Proposed Reaction Mechanism of the Rearrangement



the ketene independently prepared from carboxylic acid 5 spontaneously undergoes rearrangement to afford 2t in 85% (eq 3).



In summary, we have described development of highly efficient synthesis of 2*H*-azirine-2-carboxylic esters with quaternary centers and tandem synthesis of pyrroles. Novel rearrangement of α -oximino ketenes generated from α -diazo oxime ethers via the Wolff rearrangement results in formation of 2*H*-azirine-2-carboxylic esters in excellent yields. This reaction allowed us to further develop a tandem reaction for highly substituted pyrroles via cascade rearrangement of α -diazo oxime ethers.

ASSOCIATED CONTENT

S Supporting Information

Experimental procedures and characterization data for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

cmpark@ntu.edu.sg

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This paper is dedicated to Professor Paul A. Wender. We gratefully acknowledge Nanyang Technological University Start Up Grant and Tier 1 Grant RG 25/09 for the funding of this research.

REFERENCES

(1) Comprehensive Heterocyclic Chemistry II; Fan, W. Q.; Katritzky, A. R.; Rees, C. W.; Scriven, E. F. V., Eds.; Pergamon Press: Oxford, 1996; Vol. 2.

(2) (a) Fan, H.; Peng, J.; Hamann, M. T.; Hu, J.-F. Chem. Rev. 2007, 108, 264. (b) Grube, A.; Köck, M. Org. Lett. 2006, 8, 4675. (c) Fujita, M.; Nakao, Y.; Matsunaga, S.; Seiki, M.; Itoh, Y.; Yamashita, J.; van Soest, R. W. M.; Fusetani, N. J. Am. Chem. Soc. 2003, 125, 15700. (d) Fürstner, A. Angew. Chem., Int. Ed. 2003, 42, 3582. (e) Boger, D. L.; Boyce, C. W.; Labroli, M. A.; Sehon, C. A.; Jin, Q. J. Am. Chem. Soc. 1999, 121, 54.

(3) Lohray, B. B.; Lohray, V. Pure Appl. Chem. 2005, 77, 179.

(4) (a) Berlin, A.; Vercelli, B.; Zotti, G. Polym. Rev. (Philadelphia, PA, U.S.) 2008, 48, 493. (b) Gabriel, S.; Cécius, M.; Fleury-Frenette, K.; Cossement, D.; Hecq, M.; Ruth, N.; Jérôme, R.; Jérôme, C. Chem. Mater. 2007, 19, 2364. (c) Novák, P.; Müller, K.; Santhanam, K. S. V.; Haas, O. Chem. Rev. 1997, 97, 207.

(5) (a) Padwa, A.; Pearson, W. H. Synthetic Applications of 1,3-Dipolar Cycloaddition Chemistry toward Heterocycles and Natural Products; Wiley: New York, 2002; Vol. 59. (b) St. Cyr, D. J.; Arndtsen, B. A. J. Am. Chem. Soc. 2007, 129, 12366.

(6) (a) Estevez, V.; Villacampa, M.; Menendez, J. C. Chem. Soc. Rev.
2010, 39, 4402. (b) Balme, G. Angew. Chem., Int. Ed. 2004, 43, 6238.
(c) Fontaine, P.; Masson, G. r.; Zhu, J. Org. Lett. 2009, 11, 1555.
(d) Bharadwaj, A. R.; Scheidt, K. A. Org. Lett. 2004, 6, 2465.
(e) Dhawan, R.; Arndtsen, B. A. J. Am. Chem. Soc. 2003, 126, 468.

(7) (a) Rakshit, S.; Patureau, F. W.; Glorius, F. J. Am. Chem. Soc. **2010**, 132, 9585. (b) Huestis, M. P.; Chan, L.; Stuart, D. R.; Fagnou, K. Angew. Chem., Int. Ed. **2011**, 50, 1338. (c) Mizuno, A.; Kusama, H.; Iwasawa, N. Angew. Chem., Int. Ed. **2009**, 48, 8318.

(8) Padwa, A.; Smolanoff, J.; Tremper, A. J. Am. Chem. Soc. **1975**, 97, 4682.

(9) (a) Skepper, C. K.; Molinski, T. F. J. Org. Chem. 2008, 73, 2592.
(b) Salomon, C. E.; Williams, D. H.; Faulkner, D. J. J. Nat. Prod. 1995, 58, 1463. (c) Molinski, T. F.; Ireland, C. M. J. Org. Chem. 1988, 53, 2103.

(10) (a) Padwa, A. In Advances in Heterocyclic Chemistry; Katritzky,
A. R., Ed.; Academic Press: Orlando, FL, 2010; Vol. 99, p 1.
(b) Gilchrist, T. L.; Alves, M. J. In Organic Azides; John Wiley & Sons Ltd.: Chichester, U.K., 2010; p 167. (c) Palacios, F.; de Retana, A. M. O.; de Marigorta, E. M.; de los Santos, J. M. Eur. J. Org. Chem. 2001, 2001, 2401. (d) Palacios, F.; de Retana, A. M. O.; de Marigorta, E. M.; Santos, J. M. d. l. Org. Prep. Proced. Int. 2002, 34, 219. (e) Singh, G. S.; D'Hooghe, M.; De Kimpe, N. Chem. Rev. 2007, 107, 2080. (f) e Melo, T. M. V. D. P.; Amd'Ar, G. Curr. Org. Synth. 2004, 1, 275.

(11) (a) Jana, S.; Clements, M. D.; Sharp, B. K.; Zheng, N. Org. Lett.
2010, 12, 3736. (b) Candito, D. A.; Lautens, M. Org. Lett. 2010, 12, 3312. (c) Padwa, A.; Stengel, T. Tetrahedron Lett. 2004, 45, 5991.

(d) Johns, B. A.; Gudmundsson, K. S.; Turner, E. M.; Allen, S. H.; Jung, D. K.; Sexton, C. J.; Boyd, F. L. Jr.; Peel, M. R. *Tetrahedron* 2003, 59, 9001. (e) Stevens, K. L.; Jung, D. K.; Alberti, M. J.; Badiang, J. G.; Peckham, G. E.; Veal, J. M.; Cheung, M.; Harris, P. A.; Chamberlain, S. D.; Peel, M. R. Org. Lett. 2005, 7, 4753. (f) Brahma, S.; Ray, J. K. J. Heterocycl. Chem. 2008, 45, 311. (g) Alves, M. J.; Gil Fortes, A.; Lemos, A.; Martins, C. Synthesis 2005, 2005, 555. (h) Timen, A. S.; Fischer, A.; Somfai, P. Chem. Commun. 2003, 1150. (i) Taber, D. F.; Tian, W. J. Am. Chem. Soc. 2006, 128, 1058. (12) Rappoport, Z.; Liebman, J. F. The Chemistry of Hydroxylamines,

Oximes and Hydroxamic Acids; Wiley: Hoboken, NJ, 2008; Vol. 2. (13) (a) Banert, K. In Organic Azides; John Wiley & Sons, Ltd: Chickestry, LLK, 2010, p. 112 (b) Siikeler Timés & Bickerry, E.

Chichester, U.K., 2010; p 113. (b) Sjöholm Timén, Å.; Risberg, E.; Somfai, P. *Tetrahedron Lett.* **2003**, *44*, 5339.

(14) (a) Kirmse, W. Eur. J. Org. Chem. 2002, 2002, 2193. (b) Tidwell, T. T. Angew. Chem., Int. Ed. 2005, 44, 5778.

(15) (a) Seki, H.; Georg, G. I. J. Am. Chem. Soc. 2010, 132, 15512.
(b) Presset, M.; Coquerel, Y.; Rodriguez, J. J. Org. Chem. 2008, 74, 415.
(c) Sarpong, R.; Su, J. T.; Stoltz, B. M. J. Am. Chem. Soc. 2003, 125, 13624.
(d) Dalton, A. M.; Zhang, Y.; Davie, C. P.; Danheiser, R. L. Org. Lett. 2002, 4, 2465.
(e) de Lucas, N. C.; Netto-Ferreira, J. C.; Andraos, J.; Scaiano, J. C. J. Org. Chem. 2001, 66, 5016.

(16) Lourdusamy, E.; Yao, L.; Park, C.-M. Angew. Chem., Int. Ed. 2010, 49, 7963.

(17) Preparation of α -diazo β -keto oxime ethers. For details, see ref 16 and the Supporting Information.



(18) (a) Manning, J. R.; Davies, H. M. L. J. Am. Chem. Soc. 2008, 130, 8602. (b) Manning, J. R.; Davies, H. M. L. Tetrahedron 2008, 64, 6901.

(19) (a) P. Marsden, S.; Pang, W.-K. Chem. Commun. 1999, 1199.
(b) Collomb, D.; Doutheau, A. Tetrahedron Lett. 1997, 38, 1397.
(c) Karady, S.; Amato, J. S.; Reamer, R. A.; Weinstock, L. M. Tetrahedron Lett. 1996, 37, 8277.

(20) Qi, X.; Jiang, Y.; Park, C.-M. Chem. Commun. 2011, 47, 7848.(21) Plausible mechanism for the formation of 2k".

